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Diels–Alder approaches to ring-functionalized cyclic β-amino acids

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Abstract

Catalytic asymmetric Diels–Alder reactions with aminodiene 1 and desymmetrized fumarate 8 were used for efficient access to dihydroxylated *cis*- and *trans*-aminocyclohexane β -amino acids. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: enantioselective Diels-Alder reaction; β -amino acids; aminocyclohexanecarboxylate; scandium triflate; Curtius rearrangement.

The use of oligomers of β -amino acids as novel foldamers that could mimic physical characteristics and biological functions of peptides has led to a considerable need for efficient synthetic routes toward the enantiopure building blocks.¹ In particular, water-soluble hydroxy-lated or aminated derivatives of cyclohexyl β -amino acids have attractive properties and a high intrinsic propensity for helical folding.^{2,3} We now report an efficient approach toward dihydroxy-lated *cis*- and *trans*-aminocyclohexanecarboxylic acids that takes advantage of modern catalytic asymmetric Diels–Alder reactions (Fig. 1).

Starting with readily available *N*-Cbz-protected aminodiene $1,^{4,5}$ we explored the use of Kobayashi's chiral scandium catalyst for the Lewis acid-mediated Diels–Alder reaction with acyl-1,3-oxazolidin-2-one 2 (Scheme 1).^{6,7} After a reaction time of 24 h, *endo* product 3 was isolated in 92% yield and in 90% ee.⁸ The corresponding *exo*-product was also identified in 6% yield in the reaction mixture. After saponification of 3 and benzyl ester formation, dihydroxylation proceeded smoothly to give the desired diol 4 in 74% overall yield from 1.

We also explored an alternative route to the β -amino acid building block 4 that took advantage of a recently reported nucleophilic catalyst by MacMillan.⁹ Addition of 20 mol% of the phenylalanine-derived imidazolidinone 7 to a mixture of aminodiene 1 and acrolein in nitromethane/water provided cyclohexene 6 as a 14:1 mixture of *endo:exo* diastereomers

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Figure 1. Catalytic asymmetric Diels-Alder approaches toward cis-aminocyclohexanecarboxylate 4



Scheme 1. Chiral Lewis acid Diels-Alder approach

(Scheme 2). The use of 15 mol% of NaHCO₃ was essential in this reaction due to the instability of the aminodiene component towards protic acid and the pH-dependence of the reaction rate. Pure **6** was isolated in 68% yield and in 89% ee after column chromatography and readily converted to **4** by oxidation, esterification and dihydroxylation with catalytic OsO₄ under the Upjohn conditions. Both the Lewis acid- and the nucleophile-catalyzed asymmetric Diels–Alder routes, thus provided the desired dihydroxylated β -amino acid **4** in high enantiomeric excess that could be further increased to >98% by crystallization of **3**.¹⁰



Scheme 2. Nucleophile-catalyzed Diels-Alder approach

While epimerization at the carbonyl α -carbon of *cis*-aminocyclohexanecarboxylates has been used for the preparation of the corresponding *trans*-isomers,³ we found that this approach was unreliable in providing preparative scale access to the desired *trans*-analogs of **4**. In contrast, a stereoselective strategy using Narasaka's chiral titanium catalyst was much more efficient (Scheme 3).¹¹ Treatment of fumarate hemi-ester **8**¹² with butadiene and 10 mol% of titanium complex in the presence of 12 mol% of chiral diol **9** led to a slow, but highly enantioselective, Diels–Alder process. If the reaction was allowed to proceed for 68 h at 0°C in the presence of 4 Å molecular sieves, a quantitative yield of **10** in 85% ee was isolated.¹³ Increasing the reaction temperature considerably accelerated the process, but was accompanied by significant reductions in enantioselectivity. A single recrystallization of **10** led to material of >98% ee.



Scheme 3. Chiral Lewis acid Diels-Alder/Curtius rearrangement approach

Cyclohexene diester **10** was selectively saponified at 0°C, and the resulting acid was subjected to a Curtius rearrangement with DPPA in *t*-butanol (Scheme 3).^{1e,14} While this strategy provided a convenient access to the desired *trans*-aminocyclohexenecarboxylate **11**, introduction of the backbone hydroxyl groups that would provide for suitable water-solubility of this β -amino acid building block was problematic. Hydroxylation under a range of reaction conditions only led to mixtures of diastereomers due to lack of facial- and regioselectivity in the attack onto the alkene moiety of **11**. In contrast, exposure of the acid derived from **11** to metachloroperbenzoic acid and in situ cyclization of the intermediate epoxy carboxylate provided the hydroxy lactone **12** as a single diastereomer.¹⁵ Subsequent mild methanolysis led to the all-*trans* tetrasubstituted cyclohexane derivative **13**.¹⁶

In conclusion, we have been able to identify three stereocontrolled and efficient Diels–Alder approaches toward dihydroxylated *cis*- and *trans*-aminocyclohexanecarboxylates. Both Lewis acid and nucleophilic catalysis are suitable for the preparation of the *cis*-derivative **4** from aminodiene **1**, and a combination of a titanium-catalyzed asymmetric Diels–Alder reaction with fumarate **8** followed by Curtius rearrangement and neighboring group-assisted alkene function-alization provides the *trans*-analog **13**. We expect that building blocks **4** and **13** will be of considerable use for the design of helical, water-soluble β -peptidic foldamers as well as for the synthesis of new chiral ligands for asymmetric catalysis, and further studies toward these goals will be reported in due course.

Acknowledgements

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The residue was purified by chromatography on SiO₂ (EtOAc/hexanes, 50:50) to give 10 as a white solid (568 mg, 97, 85% ee): mp 100.0–102.0°C; $[\alpha]_D$ –149 (c 1.50, CH₂Cl₂); ¹H NMR δ 5.68–5.67 (m, 2H), 4.44-4.38 (m, 2H), 4.08-3.92 (m, 2H), 3.64 (s, 3H), 3.06-2.96 (m, 1H), 2.55-2.40, 2.20-1.95 (2m, 5H); ¹³C NMR δ 176.1, 175.5, 153.2, 125.0, 124.9, 62.0, 51.9, 42.7, 41.2, 39.8, 28.2, 28.1. (1R,2R)-Cyclohex-4-ene-1,2-dicarboxylic acid monomethyl ester. To a solution of 10 (440 mg, 1.74 mmol) in THF (27.0 mL) and H₂O (7.6 mL) was added at 0°C 30% H₂O₂ (473 mg, 1.42 mL, 13.9 mmol)) followed by LiOH·H₂O (91.0 mg, 2.17 mmol). The resulting mixture was stirred at room temperature for 18 h, cooled to 0°C, and treated with a 1.35N solution of Na₂SO₃ in H₂O (12.5 mL) followed by 0.5N NaHCO₃ (21.8 mL). The THF was evaporated in vacuo. The aqueous residue was diluted with H₂O and extracted with CH₂Cl₂ to remove oxazolidinone. The aqueous phase was acidified to pH 1-2 with 5N HCl and extracted with EtOAc. The combined organic layers were dried (MgSO₄) and concentrated. The residue was purified by chromatography on SiO₂ (MeOH/CH₂Cl₂, 5:95) to give (1R,2R)-cyclohex-4-ene-1,2-dicarboxylic acid monomethyl ester as a colorless oil (272 mg, 85%). (1R,6R)-6-tert-Butoxycarbonylamino-cyclohex-3-ene carboxylic acid methyl ester (11). To a solution of the ester (262 mg, 1.42 mmol) in dry 'BuOH (7.5 mL) was added Et₃N (173 mg, 1.71 mmol) and diphenylphosphonyl azide (471 mg, 1.71 mmol) at room temperature. The resulting mixture was stirred at room temperature for 1 h, and then heated at reflux for 24 h. After cooling to room temperature, the solvent was removed in vacuo. The residue was purified by chromatography on SiO₂ (EtOAc/hexanes, 8:92) to give 11 as a white crystalline solid (270 mg, 74%): mp 83.6-84.2°C; [α]_D -25.1 (*c* 1.12, CH₂Cl₂); IR (neat) 3385, 3032, 2984, 2952, 2926, 1723, 1683, 1525, 1435, 1369, 1312, 1256, 1192, 1168, 1016, 660 cm⁻¹; ¹H NMR δ 5.60–5.67 (m, 2H), 4.64 (br-s, 1H), 4.01 (br-s, 1H), 3.68 (s, 1H), 3.68 3H), 2.69–2.64 (m, 1H), 2.54–2.46 (m, 2H), 2.44–2.42 (m, 1H), 1.99–1.91 (m, 1H), 1.42 (s, 9H); ¹³C NMR δ 174.1, 155.1, 125.1, 124.3, 79.4, 51.9, 47.4, 44.8, 31.2, 28.4, 26.6. (1R,6R)-6-tert-Butoxycarbonylamino-cyclohex-3-ene carboxylic acid. A solution of 11 (89.7 mg, 0.351 mmol) in THF/water (10:1, 4.4 mL) was treated with LiOH·H₂O (73.7 mg, 1.76 mmol). The resulting mixture was stirred at room temperature for 78 h before THF was removed in vacuo. The residue was diluted with water, acidified to pH 1-2 with 5N HCl, and extracted with EtOAc. The combined organic layers were dried (MgSO₄) and concentrated. The residue was purified by chromatography on SiO_2 (MeOH/CH₂Cl₂, 6:94) to give (1*R*,6*R*)-6-tert-butoxycarbonylamino-cyclohex-3-ene carboxylic acid as a colorless oil (82.7 mg, 98%). (1R,2R,4S,5S)-(4-Hydroxy-7-oxo-6-oxa-bicyclo] 3.2.1 [oct-2-yl]-carbamic acid tertbutyl ester (12). A solution of this acid (86.2 mg, 0.357 mmol) in CCl_4 (3.6 mL) was treated with *m*-chloroperoxybenzoic acid (67.8 mg, 0.393 mmol) at 0°C for 4 h. Then, triethylamine (79.5 mg, 0.786 mmol) was added. The reaction mixture was heated at 55°C for 3.5 h. The solvent was removed in vacuo. The residue was purified by chromatography on SiO₂ (ethyl acetate/hexanes, 50:50) to give slightly impure **12** as a white solid (38.6 mg, 42%) that was carried on without further purification. (1R,2R,4S,5S)-2-tert-Butoxycarbonylamino-4,5-dihydroxy-cyclohexane carboxylic acid methyl ester (13). A solution of crude 12 (34.9 mg, 0.136 mmol) in MeOH (1.4 mL) was treated with sodium bicarbonate (11.4 mg, 0.135 mmol). The reaction mixture was stirred at room temperature for 27 h. The solvent was removed in vacuo and the residue was purified by chromatography on SiO_2 (ethyl acetate/hexanes, 80:20) to give **13** as a white solid (29.7 mg, 76%): mp 155.0–155.5°C; $[\alpha]_D$ –13.8 (c 1.0, MeOH); ¹H NMR (CD₃OD) δ 6.72 (d, 1H, J=8.7 Hz), 4.86 (s, 1H), 3.63 (s, 3H), 3.37–3.24 (m, 2H), 2.45–2.37 (m, 1H), 2.09–1.96 (m, 2H), 1.59–1.28 (m, 11H); ¹³C NMR (CD₃OD) δ 173.8, 156.1, 78.7, 72.9, 72.4, 51.1, 49.5, 38.5, 33.6, 27.3; HRMS (EI) m/e calculated for C₉H₁₅O₆N (M-C₄H₈): 233.0899, found: 233.0897.